

H₂S Oxidation over Atomic Cobalt Modified Nitrogen-Rich Hierarchical Porous Carbon Nanofibers: Efficient Desulfurization and High-Performance Li-S Batteries

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ABSTRACT

Cobalt atoms modified nitrogen-rich hierarchical carbon nanofibers (CoN-MCNFs) with an efficient activity for the selective oxidation of H₂S have been constructed through the combination of electrospinning and carbonization. The doped atomic cobalt/nitrogen content, the specific surface area (S_{BET}) and the ratio of micropore to mesopore of CoN-MCNF hybrids can be readily adjusted by tuning the experimental parameters, including the proportion of Co in Zn,Co-ZIF, the ratio of PAN to Zn,Co-ZIF in precursors and the carbonization temperature, as well as the activation by KOH. Thanks to their unique hierarchical porous structures and the co-doping of Co and N, the as-obtained CoN-MCNF hybrids exhibit superior sulfur capacity (4.69 g H₂S/g cat.) and higher selectivity for the oxidation of H₂S into elemental sulfur at room temperature, compared with the original CNFs and microporous carbon derived from direct carbonization of Zn,Co-ZIF. More interestingly, owing to the novel structures and high sulfur loads, the spent CoN-MCNF catalysts can be further applied as a cathode in lithium-sulfur battery (LSB) and exhibits a high Coulombic efficiency, good rate capability, as well as preferable cycling stability. This strategy opens up a new avenue to fabricate heterogeneous nanocatalysts for the efficient removal of pollutants and the creation of novel materials for high-performance energy storage.

KEYWORDS: H₂S oxidation; Porous carbon nanofibers; Cobalt and Nitrogen co-doping; Electrospinning method